

Total number of printed pages-8

3 (Sem-6/CBCS) CHE HC 2

2023

CHEMISTRY

(Honours Core)

Paper : CHE-HC-6026

(Organic Chemistry-V)

Full Marks : 60

Time : Three hours

***The figures in the margin indicate
full marks for the questions.***

1. Answer the following questions : $1 \times 7 = 7$

(a) What do you mean by fingerprint region ?

(b) Which of the following is a chromophore ?

(i) $-\text{SO}_3\text{H}$

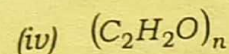
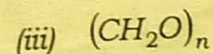
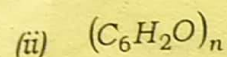
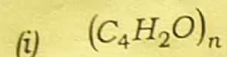
(ii) $-\text{OH}$

(iii) $-\text{COOH}$

(iv) $-\text{NO}_2$

Contd.

(c) The general formula of carbohydrate is—



(d) Which of the following compounds do not absorb light above $200m\mu$?

(i) Ethanol

(ii) Diethyl ether

(iii) 2-Butanone

(iv) Benzene

(e) At what wavelength range the coloured compounds absorb?

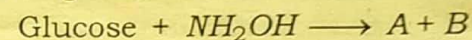
(f) Give *one* example of a thermosetting plastic.

(g) What are the expected products of hydrolysis of Lactose?

2. Give answer of the following : $2 \times 4 = 8$

(a) What are the different types of electronic transitions that occur in an organic molecule?

(b) Find out the products *A* and *B* in the following reaction :



(c) What is a mordant dye? Give *one* example. $1+1=2$

(d) Give *one* example of each of the following polymers : $1+1=2$

(i) Polyamides

(ii) Polyesters

3. Answer **any three** of the following :

$5 \times 3 = 15$

(a) (i) How can you distinguish between intra and inter-molecular hydrogen bonding with the help of IR spectroscopy? 3

(ii) How can you distinguish the following pair of compounds using IR spectroscopy Propanone and Propanal? 2

(b) Fructose contains a keto group, but still it gives silver mirror test on treatment with Tollen's reagent. Explain by showing the rearrangement reactions involved. What is the name of the rearrangement reaction? $2+2+1=5$

(c) Match the following in 'A' with those given in 'B' : $1 \times 5 = 5$

'A'

'B'

- | | |
|----------------------|----------------------|
| (i) D-Sorbitol | (a) Anomeric carbon |
| (ii) L-Ascorbic acid | (b) A disaccharide |
| (iii) Glycoside | (c) A sugar lactone |
| (iv) C-L of glucose | (d) Sugar alcohol |
| (v) Maltose | (e) A reducing sugar |

(d) Write the synthesis of Congo red dye. Show the structural changes involved due to which it changes color from red to blue in acid solution. $2+3=5$

(e) (i) Write the full form of the following terms : 2

- (i) PAN
- (ii) PTFE
- (iii) PCTFE
- (iv) BSR

(ii) What are polyolefins and polydienes? Give *one* example of each. $1+2=3$

4. Answer **any three** of the following :

$10 \times 3 = 30$

(a) (i) Draw the Fisher's Projection formula of *D*-glucose.

(ii) What do you mean by the term anomerization? Show the mechanism of anomerization of *D*-glucose.

(iii) Draw the chair conformers of both the α -*D*(+) and β -*D*(+) glucopyranose.

(iv) How do you explain the greater stability of β -*D*(+)-glucopyranose from their conformers?

$1+3+3+3=10$

(b) (i) How will you bring the following conversions? $3 \times 2 = 6$

(A) An aldopentose to an aldohexose

(B) *D*-fructose to *D*-glucose

- (iii) Predict the chemical shift positions for the protons in 1-bromoethane and hence draw a rough sketch of the ^1H NMR spectrum. 3
- (iv) In a ^1H NMR spectrum, the protons of ethene appear at a more downfield region than expected. Why? 2
- (f) (i) Explain the basic principle of NMR spectroscopy. 5
- (ii) Explain, why ESR spectrum is recorded in derivative mode? 2
- (iii) Predict and draw the hyperfine structure of CH_3 using ESR spectroscopy. 3